



# Chemistry 111, General Chemistry Laboratory A

## Spring 2023 Syllabus

**The following syllabus applies to all of the *in-person* lab sections: Chem 111-001 to 111-014.  
Lab Location: Flanner Hall 305-308**

Students can only attend the section in which they are enrolled in LOCUS. Be mindful of day/time and location. All days and times are central standard time (CST).

Pre/Co-requisite: Chem 101

Prerequisite: Math Placement Test or Math 117

Laboratory Coordinator: Dr. Andrew Basner

Office Hours: Wednesday, 12:00-2:00pm Office Hours Location: STEM Center is in St. Joseph's Hall cafeteria.

Office Location: Room 428 Flanner Hall | Office Phone: 773-508-3135

Email: [abasner@luc.edu](mailto:abasner@luc.edu)

Teaching Assistants (TAs) will be assisting the lab coordinators during this course. Specific TA information is listed in Sakai, along with the Primary Lab Coordinator for your section.

Welcome to Chem 111. We are looking forward to working with you this semester. Read the entire syllabus to understand the expectations of this course.

## **COURSE DESCRIPTION**

This lab course emphasizes introductory application of topics/theory covered in the lecture course (Chem 101). It introduces students to basic chemical laboratory skills & techniques including lab safety, accuracy/precision, net ionic equations, pH, enthalpy, spectrophotometry, use indicators, a variety of laboratory equipment & glassware, basic statistics, etc. This list is not exhaustive but mentions the highlights.

Goals of course:

- 1) Learn lab safety
- 2) Learn basic laboratory skills
- 3) Apply theory to practical use

Outcomes:

- 1) Students will know proper safe lab practices
- 2) Students will be able to properly identify & use glassware, analytical balances, hot plates, and other lab equipment based on the task at hand
- 3) Students will be able to compute calculations and theoretical analysis questions coupled to each lab experiment.

## **REQUIRED ITEMS**

- 1) Chem 111 Laboratory Packet (provided as a PDF in Sakai). Printed manuals stored in the lab are provided during synchronous session. Feel free to print out a personal copy of the manual if you prefer to.
- 2) Chem 111 Labster Lab Manual (provided as a PDF). Simulations are done online, so this manual is not printed. Feel free to print a personal copy if you want to. Printed copies will not be provided to students.
- 3) Labster – web-based virtual lab experiment simulations to learn some lab techniques. Access to Labster will be provided in Sakai for free to students!
- 4) Composition style notebook (not spiral bound & cannot have tear-out perforations). Line ruled.
- 5) A non-erasable pen is required for all written work. No white-out is allowed. No pencils allowed.
- 6) Safety goggles (we provide to you). These must be type G, H or K goggles and must meet or exceed ANSI Z87.1 requirements. Safety glasses do not meet our requirements and are not allowed.
- 7) Long-sleeve Laboratory Coat (white is preferred coat color). You must purchase this.
- 8) Appropriate clothing and footwear. See below for details\*
- 9) Scientific calculator (non-programmable, non-graphing, and independent of another device such as a phone or tablet). Cell phones are not calculators.

- 10) Sakai access via the internet to watch pre-lab video content, post-lab content, online content, lab simulations, and complete all Sakai work.
- 11) Desktop or Laptop computer with internet access. Labster does NOT work on tablets nor mobile devices; Sakai does not display well on those devices. If you do not have a computer, the Information Commons (IC) on campus has plenty of available computer stations to do work at. You may also read about the [extended loan equipment program](#) within the first week of class to arrange a resource. Lab Coordinator is not responsible for coordinating resources for students nor responsible for a loaned device. Many coursework items require a computer to access and complete them.
- 12) Cam Scanner app OR a scanner machine. Cam scanner is a free app that converts a phone picture to a PDF file. You will take pictures of the Composition lab notebook pages and convert them to a PDF file in order to submit the notebook pages to Sakai for grading. This app works on android and iPhone. A scanner machine can be used as well.
- 13) Microsoft 365 (free for LUC students) to write a formal lab report. Information is supplied on [how to download & access Microsoft 365 for free](#).
- 14) Panopto (free for LUC students). One format of recorded course content is Panopto video. You may be prompted to log in with UVID username and password to view the videos. Links to videos will be provided in Sakai (Panopto tool) and via email.
- 15) Periodic table. Here is a cool one provided by the [Museum of Science & Industry](#)
- 16) (optional) A face mask. See Fall 2022 Masking Requirement below.

**\*Appropriate clothing must be worn that minimizes potential chemical contact with your skin. Shoes that adequately cover the entire foot are required. Sandals, open-toes shoes, perforated shoes, open-backed shoes are not acceptable. No skin should be exposed on your feet or legs, so clothing that covers and protects your body from the waist down (including your ankles) must be worn. Lab coat required.**

## **GENERAL POLICIES**

- Attendance is mandatory. All written & Sakai work, as well as TA observations, serves as the basis for earning points and showing progress. Written work will be graded with an emphasis on correct significant digits, consistent results (do data & observations match conclusions), appropriateness/correctness of analysis, and thoroughness in responses. Following directions of reporting calculated answers are taken into account in grades too.
- All work must be completed in non-erasable pen. This includes the Composition notebook, any worksheets/handouts and homework assignments. Work not completed in pen or containing “white-out” is subject to a point deduction and is not eligible for any regrade requests.

- The Composition notebook is to contain all laboratory experiment information [Date, Title, Introduction, data/observations/calculations, and lab experiment conclusions]. It is the major component of your course grade. Losing this Composition notebook may result in failure of the course, as one cannot be graded on work that does not exist if the notebook is lost. Follow the guidelines given on how to appropriately set-up a laboratory notebook.
- Aspects of lab work must be done in the Composition notebook only. (No loose-leaf paper or other notebooks). Your TA will evaluate/grade your notebook. This functions as a way for TAs to grade along the way and to make sure you are on the right track with lab results. If there are any discrepancies in recorded grades, proof of having earned a specific grade on a particular lab is the presence of that graded work in your notebook.
- Each student is assigned a drawer with glassware and equipment. At the beginning of the semester & semester's end, the drawer contents will be checked for completeness. The drawer is shared with other students over the course of a week. Therefore, it is essential that you clean the equipment used after an experiment is done. Drawers may be checked sporadically. If glassware is broken, the student is responsible for requesting a replacement item; there is no penalty for broken glassware.
- Homework can never be submitted via email. No exceptions.
- In general, Sakai work cannot be made up. Late work is not accepted for quizzes. Late lab notebooks, lab report, and Labsters may be accepted (**See Late Policies section**).
- Safety and Clean-up points will be earned on the basis of safe/professional conduct in the lab. A safe lab environment is essential. Unsafe actions will definitely result in grade degradation. The following is a partial list of ways you can lose safety/clean-up points:
  - Coming late to class, after the pre-lab lecture has started will result in deduction of safety points.
  - Not dressing appropriately for lab. Proper footwear/clothing are required.
  - Not bringing goggles to lab/not wearing your goggles consistently in lab can result in expulsion from the lab. Safety glasses do not meet our safety requirements.
  - Not properly wearing a face mask.
  - Not bringing a lab coat to lab. Not wearing the lab coat properly [buttoned] during lab.
  - Not keeping your equipment drawer or lab space in good condition (i.e. dirty glassware/bench).
  - Engaging in horseplay/actions that may endanger you, your classmates, TA, or Lab Coordinator.
  - Not adhering to Disposal Instructions indicated in each lab handout.
  - The lab-pro equipment used is breakable and requires special care. You and your partner will be assigned a box to use, and if the equipment is found to have been handled negligently, points will be deducted from both your safety points and your lab score for the both of you.

\*An action, even if not herein, deemed unsafe by TA or Lab Coordinator will result in safety point deductions.\*

**Failure to adhere to lab safety rules can result in expulsion from the lab session and/or course with no opportunity for make-up of the work. Safety must be taken very seriously.**

## **ATTENDANCE/PARTICIPATION**

**Attendance is mandatory.** You are required to come to class on your assigned “in-person days” and can only attend the lab section in which you are officially enrolled in LOCUS. There is a point value associated with the work accomplished in each class, and you cannot earn points for classes that you do not attend. **There are limited opportunities to make up missed labs!** A student may attend class during the asynchronous week to complete the make up in most instances. There are six lab experiments and students are expected to complete all of them. If a student is absent for both the lab and make up lab, a zero (0) is recorded in the gradebook for the experimental work missed. Sample data is given so the student can complete the homework questions pertaining to the lab missed.

Students are not allowed to make up any of the Sakai quiz work. Students are typically not allowed to make up a lab experiment in another section of Chemistry 111 due to space restraints. If the university is open, you are expected to attend class and be on time. Points are deducted for those who arrive late. If you arrive after the conclusion of the pre-lab lecture, you will not be allowed to perform the lab. Being sent home for improper clothing/footwear also counts as an absence and no makeup work is allowed.

Review the schedule at the end of the syllabus and consider the negative impact that missing a hands-on laboratory session will have on your educational experience. It is in your best interest to register for a section that does not conflict with other obligations. Students should not enroll in a lab section that they cannot fully attend. Missing 2 of the labs, which is nearly 33% of the lab work, is significant and unacceptable and will result in academic failure. The same is true for not turning in the assigned formal lab report.

### **Loyola University Absence Policy for Students in Co-Curricular Activities (including ROTC):**

Students missing classes while representing Loyola University Chicago in an official capacity (e.g. intercollegiate athletics, debate team, model government organization) shall be allowed by the faculty member of record to make up any assignments and to receive notes or other written information distributed in the missed classes.

Students should discuss with faculty the potential consequences of missing lectures and the ways in which they can be remedied. Students must provide their instructors with proper documentation (develop standard form on web) describing the reason for and date of the absence.

This documentation must be signed by an appropriate faculty or staff member, and it must be provided as far in advance of the absence as possible. It is the responsibility of the student to make up any assignments. <https://www.luc.edu/athletheadvising/attendance.shtml>

Students participating in co-curricular activities must make information concerning time conflicts with University sponsored events available to the Laboratory Coordinator within the first two weeks of the semester. The Laboratory Coordinator reserves the right to contact the Athletics Department confirming time conflicts and regarding concerns. Students missing classes while representing Loyola University Chicago in an official capacity (e.g., intercollegiate athletics, debate team, model government organization) will need to discuss their needs with the Laboratory Coordinator. Sakai work cannot be made up in any circumstances, no exceptions. Laboratory work cannot be made up either; you cannot attend another lab section. These types of absences are handled on a case-by-case basis with remedy.

### **Accommodations for religious reasons:**

Students missing a lab experiment due to observing religious holidays must alert the Lab Coordinator no later than two weeks after the start of the semester to request a special accommodation. This is handled on a case-by-case basis. The Lab Coordinator reserves the right to contact Campus Ministry, which keeps information on a plethora of religions and holidays.

Students must discuss with the Lab Coordinator the consequences of missing laboratory and the ways [if any] they can be remedied, while also providing the Laboratory Coordinator with proper documentation describing the reason and date of the absence. The document must be signed by an appropriate Faculty/Staff member, and it must be provided as far in advance of the absence as possible. It is a student's responsibility to proactively ask what will be missed due to absence.

## **COURSE REPEAT RULE**

Effective as of the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include withdrawals (W). After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to register form or print it from the Department of Chemistry & Biochemistry [website](#) and personally meet and obtain a signature from either the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt.

## **ROLE OF TEACHING ASSISTANTS**

In each lab session, your primary interaction could be with a Teaching Assistant. The function of a TA is to help you safely get good data and provide individual help when necessary. The role of the Laboratory Coordinator is more behind the scenes: plan the curriculum, prepare handouts and PowerPoints, and train the TAs so the lab experience is educational, fair, and effectively run for students enrolled in all the sections. Know that the Laboratory Coordinator will be in lab too but may step outside of the lab from time to time to handle appropriate curriculum work. The Laboratory Coordinator is available to you during and outside of the laboratory hours if there are

any questions or concerns that the TAs cannot handle appropriately. The Laboratory Coordinator has final authority in all matters relating to the course. TAs will keep office hours, which will be posted on Sakai. Utilize both your Lab Coordinator and TA if you need assistance. \*If at any point during the semester you want to talk to a Laboratory Coordinator regarding your TA, please do. The TA should enhance your educational experience. If this is not the case, talk to a Lab Coordinator so they know.

## **TUTORING**

The Tutoring Center offers free tutoring. Visit the [Tutoring Center Page](#).

## **GRADING**

Reference the grading scale below. There will be no change in the grading scale nor the number of points allotted in this course. It is in your benefit to attend all 6 in person lab experiments to learn the content for homework, quiz, and/or lab report. There are no weighted grades in the course; it is points earned divided by total points possible. A zero (0) is earned for work not completed. If you do not bring your lab notebook to lab, points will be deducted.

The University uses the +/- grading scale system and that system is implemented in this course. Rounding only applies to the final course grade percentage. Sakai reports course grades to TWO digits past the decimal (XX.XX%); this final course grade percentage is rounded to the closest integer. For example, an 89.50% or 89.90% (B+) rounds up to a 90% (A-), BUT an 89.30% or 89.45% (B+) round to the integer 89% (B+).

Grades are posted on Sakai within one week of completing the work [Excel homework, notebook entries, Sakai pre/post work]. Any grading discrepancies must be resolved the day the graded work is handed back. Discrepancies in Sakai [grade incorrect, for example] must be resolved no later than one week after reviewing the graded assignment. A student must show proof the work was graded wrong or in the gradebook incorrectly. Grade disputes will not be entertained past 1-week of the graded work being returned to students nor be acknowledged the last day of class. Be mindful of this policy. Efforts are made to ensure that all Chem 111 TAs uniformly grade. On very rare occasions, if it is found there are differences between TA's grading a lab section's averages may be scaled to adjust. When this is necessary, the average lab score mean is adjusted to the average quiz mean.

**Grading Scale:**

% total	Grade
94 – 100	A
90 – 93	A-
87 – 89	B+
84 – 86	B
80 – 83	B-
77 – 79	C+
74 – 76	C
70 – 73	C-
65 – 69	D+
60 – 64	D
0 – 59	F

**Point Breakdown:**

Activity	Origin	Points	% of Final Grade
Labster - Laboratory Safety virtual lab simulation	Labster	110	<b>10%</b>
Labster - Chemistry Safety virtual lab simulation	Labster	130	
Labster - Solution Preparation: From Salt to Solution virtual lab simulation	Labster	60	
Labster - Titration: Neutralize Acid Lake Contamination virtual lab simulation	Labster	120	
Labster - Pipetting: Master the Technique virtual lab simulation	Labster	140	
Labster - Spectrophotometers: Building and Exploring the Instrument virtual lab simulation	Labster	30	
<b>Total Labster Points</b>	<b>Labster</b>	<b>590</b>	
Notebook Pages OR Data Sheet – Beanium	In Lab/Sakai (Assignments)	20	<b>45%</b>
Notebook Pages – Volumetric Measurements	In Lab/Sakai (Assignments)	20	
Notebook Pages - Determination of Calcium concentration in Loyola Water	In Lab/Sakai (Assignments)	20	
Notebook Pages - Analysis of Vitamin C content in a Supplement Tablet	In Lab/Sakai (Assignments)	20	
Notebook Pages - %H <sub>2</sub> O <sub>2</sub> in Dental Whiteners	In Lab/Sakai (Assignments)	20	



Notebook Pages - Spectrophotometric Analysis of a Sports Drink	In Lab/Sakai (Assignments)	20	
Notebook Pages - Spectrophotometric Analysis of Aspirin Tablet	In Lab/Sakai (Assignments)	20	
<b>Total Lab + Notebook Points</b>	<b>Sakai (Assignments)</b>	<b>140</b>	
Quiz 1 – Syllabus, Course Policies, and Safety	Sakai (Tests & Quizzes)	20	<b>20%</b>
Quiz 2 - Beanium	Sakai (Tests & Quizzes)	20	
Quiz 3 - Determination of Calcium concentration in Loyola Water	Sakai (Tests & Quizzes)	20	
Quiz 4 – Titrations and Stoichiometry	Sakai (Tests & Quizzes)	20	
Quiz 5 - Spectrophotometric Analysis	Sakai (Tests & Quizzes)	20	
Quiz 6 – Cumulative Quiz	Sakai (Tests & Quizzes)	20	
<b>Total Quiz Points (Lowest Dropped)</b>	<b>(Lowest Quiz Dropped)</b>	<b>100</b>	
Excel - Basic Statistics in Excel	Sakai (Assignments)	20	<b>5%</b>
Excel - Graphing in Excel, Sports Drink Lab	Sakai (Assignments)	20	
Excel - Graphing in Excel, Aspirin Tablet Lab	Sakai (Assignments)	20	
<b>Total Excel Activities Points</b>		<b>60</b>	
Formal Laboratory Report 1, Peer Review	Sakai (Assignments)	15	<b>15%</b>
Formal Laboratory Report 1, Final Submission	Sakai (Assignments)	85	
<b>Total Formal Laboratory Report Points</b>		<b>100</b>	
Clean Up, In Lab Safety, Lab Prep	In Lab (Labs #1-7)	10/lab	<b>5%</b>
<b>Total Clean Up, In Lab Safety, Lab Prep</b>		<b>70</b>	

### **Grade if Absent:**

A zero (0) is recorded for work not completed, absent or not. This score will be updated after the required work is completed and submitted if allowed - see late work policies below. The Lab Coordinator also has the right to fail the student if two or more absences occur during the semester or if the midterm and/or lab report is not turned in.

**If you miss a lab, contact your primary Laboratory Coordinator immediately.** Request sample data for the lab experiment missed. The sample data is similar to what you may have obtained in lab and may help you study for homework and/or a lab report. You will be responsible for understanding the missed material, and **normal deadlines always apply for completing homework on Sakai.**

### **Late Work Policies:**

**QUIZZES:** If not completed on time, a 0 is the final grade. Quizzes cannot be accessed after the due date.

**NOTEBOOK ENTRIES:** If not completed on time, a 1-week grace period is allotted to turn the work in late [2pt penalty for lateness applied to grade]. After 1-week, if the notebook is not turned in, a 0 is the final grade.

**EXCEL WORK:** If not completed on time, a 1-week grace period is allotted to turn the work in late [2pt penalty for lateness applied to grade]. After 1-week, if the Excel work is not turned in, a 0 is the final grade.

**FORMAL LAB REPORT 1:** If the first draft is not completed on time, a 24-hr grace period is allotted because there is a peer review involved. If draft is not turned in, student forfeits 15pts allotted to peer review [since a peer's review cannot be done if a student doesn't turn in a draft of their own typed formal lab report]. For the final draft of the report, a 1-week grace period is allotted to turn the work in late [2pt penalty per day for lateness applied to grade]. After 1-week, if the final draft of the typed formal lab report is not turned in a 0 is the final grade.

**LABSTERS:** If not completed on time, a 1-week grace period is allotted to turn the work in late. After 1-week late, if the Labster work is not turned in a 0 is the final grade.

As a student in this course, by enrolling in this online course you agree to follow and abide by all due dates and understand grades will suffer if work is not turned in on time or if work is not turned in at all.

There is no final exam in this course.

## **EDUCATIONAL GOAL**

In this general chemistry laboratory course, my purpose as your Chemistry Instructor / Lab Coordinator is to provide a hands-on introduction to experimental methods of scientific investigation in Chemistry. The fundamental models of chemistry discussed in lecture will provide the basis for understanding the experimental laboratory work. Each lab will provide a practical opportunity for you to gain competence with the basic techniques of lab work and the practical experience necessary to understand its significance. It is my wish that this laboratory experience will encourage students who are seeking intellectual challenges along with an understanding of the chemical principles in the laboratory.

Conducting experiments and collecting data to test the validity of theories and models requires a different set of skills than those required for success in the lecture part of a general chemistry course. During a laboratory activity, each student's hands, mind, eyes, as well as other senses are focused on the task at hand. Success in the lab involves skills in making perceptive qualitative observations and accurate quantitative measurements.

With each laboratory experiment, relevant questions are posed, and along with TAs, I help each student to execute a laboratory approach which will yield reliable data related to these questions. Each student is required to obtain data and to depend upon this data when answers to these questions are drafted. All labs are structured enough so that you should not feel lost or confused, but not so structured that you will find it unnecessary to think for yourself.

## **PRIVACY POLICY AND SHARING OF COURSE MATERIALS**

Materials from the course cannot be shared outside the course without the instructor's Written permission.

## **REGARDING SAKAI AND TECHNICAL DIFFICULTIES**

It is *strongly encouraged* that all required submissions to Sakai as well as writing lab reports, opening course/data/experiment files, be done on a reliable wired internet connection [not wireless], that of which the University itself provides in the Information Commons and various computer labs on the Lake Shore Campus. Under NO circumstances will excuses of "technical difficulties" be accepted as this syllabus is stating all students should use a wired internet University computer [not wireless internet] to submit work in Sakai, write lab reports, open course/data/experiment files. This list is not exhaustive and it should be noted that any activities this course may require a computer or internet connection for should be completed using University computers with wired internet connection. Use of home internet [wired or wireless], University wireless, or public wireless is at your, the student's, own risk. It is not prohibited but as Instructor has stated in this syllabus, Instructor is not responsible for technical difficulties of non-University devices [cell phone, tablet, home/work/public wireless internet or computer]. Do not submit items in Sakai using a cell phone or a tablet device as these do not count as reliable internet connection tools."

## **ACADEMIC INTEGRITY**

The standard of academic integrity and personal honesty delineated in the College of Arts & Sciences [Statement on Academic Integrity is expected of every student and will be enforced.](#) Cheating can take many forms in lab, but the most common forms are copying data and answers to analysis questions, sharing files for homework, or completing Sakai work with another person. The data and analysis as well as the homework submitted for grading must be your own. If it is not, no credit will be awarded for the entire lab, nor will make-ups be granted. Findings of dishonest academic behavior are reported to the Chair of the Chemistry Department and to the Dean's Office, and are entered into an individual's record. Copied answer/report will result in penalty for all students involved.

## **DISABILITY ACCOMMODATIONS**

If you have a documented disability and wish to discuss academic accommodations, see your primary Laboratory Coordinator by the second meeting of lab. The Coordinator of Student Accessibility Center (SAC), formerly referred to as SSWD, is located in the Sullivan Center and must be contacted independently.

Necessary accommodations will be made for students with disabilities who procure a SAC letter. However, to receive any accommodations self-disclosure, proper documentation, and registration with the SAC office at Loyola University Chicago is required. Accommodations cannot be made until the Laboratory Coordinator receives proper documentation. Furthermore, accommodations are not retro-active and begin only once appropriate documentation has been received by the Laboratory Coordinator in a timely manner. Recognize that the course time scheduled in LOCUS is fixed. No extra time on wet chemistry is given to a student with an SAC letter; it is not possible and the SAC office has been made aware of this.

Only those accommodations that are specifically listed in the formal SAC letter will be provided. If an accommodation letter suggests the Testing Center be utilized to take an exam, it is the student's responsibility to schedule the testing time in the center. Note there are no exams in this course. Review the [SAC Policies and Procedures](#).

## **SMART EVALS**

Feedback on the course is important so that a Lab Coordinator can gain insight into how to improve the course, the teaching style, and so the department can learn how best to shape the curriculum for future semesters. Towards the end of the semester, students will receive an email from the Office of Institutional Effectiveness with a reminder to provide feedback on the Chem 111 course the student is enrolled in. This office will send you constant reminders during the open period of feedback until the evaluation has been completed.

The evaluation is completely anonymous. When the results are released, no one will be able to tell which student provided the individual feedback. The feedback is not released until after the semester is over, therefore any feedback given will not impact student grades.

## **LOST AND FOUND**

Any items mistakenly left in lab will be taken to the Chemistry Department office, 125 Flanner Hall, and can be identified and claimed there. **Please put your name on your Composition notebook, lab manual, calculators, lab goggles, lab coat [tag], and other personal items.**

## **ADDITIONAL STUDENT RESOURCES**

Below are links to information in the event that students need more structured guidance on using the tools in the course. A link to the University Help Desk is provided for technology questions. Students can email the Lab Coordinator. However, links below may reveal the answer quickly.

[First and Second Year Advising](#)

[Information Technology Service Desk](#) (ITS Help Desk)

[Labster Simulation Support and Tips](#)

[Panopto Information](#)

[Resource Guide for Online Learning](#)

[SAKAI student guide](#)

[Success Coaching and Writing Center](#)

[Student Accessibility Center](#)

[ZOOM Information](#) and [Contacting ZOOM Support](#)

## **SAFETY IN THE LABORATORY**

**Laboratory safety is everyone's responsibility. By registering for and participating in this course you agree to abide by all of the safety precautions, information, and rules provided to you herein as well as in or outside of the laboratory. Failure to follow these rules constitutes grounds for withdrawing the offending student from the lab session and or course at any time.**

The Laboratory Coordinator, TA, and University take safety in the laboratory very seriously. Make sure to always listen to information regarding extra safety precautions when applicable. The rules of safety listed on the following page are reviewed during the first day of the

laboratory course. Practice safe laboratory conduct during the entire semester and beyond. This list is not exhaustive and it is the student's responsibility to understand the proper, safe conduct when working in a laboratory. Students cannot complete experiments in the course unless the safety lecture and safety form are completed.

By using common sense and following all the safety rules provided, it is unlikely that you or your classmates will be involved in or injured in a mishap in the laboratory. While it is very important that you do your part to prevent an accident from occurring, it is just as important to know what to do if someone is injured.

There are several key safety features of a laboratory that will be pointed out the first day of class.

Preventing an accident or injury from occur is the ideal case scenario, which is why proactively being safe in the laboratory is desired. We live in the real-world and therefore have to be reactive in case of a lab incident.

Although not a requirement, it can be very helpful if a Laboratory Coordinator knows if a student has a condition that could possibly render an unsafe lab situation (allergies to latex, heart condition, seizure risk, etc.). Do feel free to discuss any concerns you may have regarding health conditions and laboratory work.

**Your commitment to safety [including the following rules] is very important:**

1. To always be on time to lab. Coming in late violates safety. Pre-lab lecture starts on time and missing any of its content is unsafe.
2. To wear approved safety goggles<sup>1</sup> and a [buttoned] long-sleeve laboratory coat at all times in the laboratory. Safety glasses are NOT allowed under any circumstances.
3. Non-latex, nitrile glove are optional but *highly* encouraged, especially when working with acids and bases or solvents. Do not wear gloves in the hallway or anywhere outside of lab.
4. To know both the location of and how to use eye washes.
5. Not to wear contacts in the laboratory. Eyeglasses are recommended.
6. To wear appropriate clothing that minimizes potential chemical contact with your skin. A lab coat is required, as are shoes that adequately cover the entire foot. Sandals, open-toe shoes, perforated shoes, open-backed shoes are not acceptable. No skin should be exposed on your feet, ankles, or legs, so clothing that covers and protects your body from the waist down (including ankles) should be worn. You must be dressed appropriately to do experiments.
7. To know both the location of and how to use the safety showers.
8. To know both the location of and how to use the fire extinguishers.
9. To know the proper clean-up and disposal procedure for broken glass.
10. Not to perform unauthorized and unknown experiments, nor work in the lab alone.

11. Not to take chemicals or equipment out of the laboratory.
12. Not to engage in horseplay or any clowning around that may endanger you or other students.
13. Not to eat, drink, chew gum, or smoke anything in the laboratory at any time. No headsets, cell phones, or any other audio devices.
14. Cell phones cannot be used as calculators.
15. To pull long hair back, keeping it away from chemicals and open flame.
16. To keep your lab space clean and tidy. This includes locking your lab locker when done.
17. To ask your Instructor or TA when in doubt about procedures.
18. Inform your Instructor of any health condition you have that might affect your performance or safety in the laboratory.

This list is not exhaustive. The Lab Coordinator and/or Teaching Assistants reserve the right to make a judgement call on an activity they deem unsafe taking place in the laboratory. Safety is a priority and students who do not follow the rules can be removed from the course, and if necessary Campus Safety will be called.

If you have any questions regarding the content of this syllabus, including the safety information provided, you are encouraged to discuss all questions/concerns with the Laboratory Coordinator.

The information provided on the following page are some basic reactive procedures to difference scenarios that have occurred in the laboratory.

## **FIRST AID BASICS**

*Minor Cuts:* Clean the wound, remove foreign material. Band-Aids are available. Two Band-Aid rule:

If you bleed through one Band-Aid, another should be applied over the first Band-Aid. If you bleed through two Band-Aids in a few minutes, you will be escorted to Health Services). Additionally, if there is any possibility of broken glass in a cut, you will be escorted to the Wellness Center.

*Minor Burns from Fire:* Immerse affected area in ice water.

*Chemicals in Eyes:* Immediately flush eyes with water at the eye wash. Continue with flush for at least 10 minutes. Hold the affected eye(s) open to do this properly.

*Chemicals on Skin:* Rinse affected area with water immediately at the sink or safety shower. If clothing is affected, remove clothes before rinsing! Continue to rinse for at least 10 minutes.

**Critical Injuries** may include: glass in his/her eye(s), serious cuts, severe chemical burns, severe fire burns, seizures. **Immediately call for help using either the lab phone (security number is taped to phone handle) or the emergency phone in the hallway directly**

**outside the laboratory.** Anyone with chemicals or foreign objects in his/her eye(s) will be escorted to the Wellness Center or to the hospital.

## **FIRE HAZARDS**

The primary heat source in this laboratory is the Bunsen burner, which is fueled by natural gas. A lit Bunsen burner is a small, controllable fire, but the heat generated by the burner fire can be quite hazardous in certain circumstances. It can serve as an ignition source for other combustible materials in the lab, such as paper (lab handouts, paper towels, filter paper, etc.), plastics (wash bottle), flammable liquids (acetone, ethanol). A burner fire can also ignite clothing and hair. Proper operation of a burner and the absence of combustible materials in the proximity of the burner will significantly reduce the risk of a fire.

Keep chords and paper products away from laboratory hotplates. Always make sure hot plates are off & un-plugged before leaving the lab. Avoid spilling chemicals on hot plates.

Each lab is equipped with several fire extinguishers, fire blanket, and safety showers, which should be used in a fire emergency.

### *In a case of a fire:*

Remain calm; alert the instructor and your immediate neighbors.

Personal safety, yours and others in the labs, is always the top priority.

A small fire in a small container can be suffocated by covering it with a watch glass or inverted beaker.

With a somewhat larger fire, decide whether or not you think you can control it with a fire extinguisher.

### *Use of a Fire Extinguisher:*

Located by the doors in both labs; a back-up fire extinguisher is located at the west end of the floor.

Maintain an escape position; i.e. stay between the fire and the doorway.

**Break the plastic ring, pull out the metal ring, release the hose from the bracket, direct the hose at the base of the flames, and press the lever down. PASS (pull, aim, squeeze, sweep).**

Note: Fire extinguishers are heavy and not particularly easy to direct. These are multi-purpose, dry chemical extinguishers, safe for anything we use in lab.

The Laboratory Coordinators reserve the right to revise this syllabus to correct any unintentional mistakes and/or to change the labs or directions for the class if necessary. Students will be notified if any changes have been made.

## **SPRING 2023 MASKING REQUIREMENT**

For more information on the current mask policy, please see the [University's Health, Safety Website](#). A mask may be required depending on the preference of the lab coordinator and you may choose to wear one at your own discretion based on your comfort level.



**Academic Calendar:** It is the student's responsibility to know the schedule for this course as well as the official [University Academic Calendar](#) of important dates.

### **Tentative Chem 111 Order of Lab Experiments**

Following the first lab, students will be grouped into two groups, A and B. These groups will alternate between synchronous and asynchronous activities depending on the week OR will both be in person. The tentative order of activities for each group is shown below.

Safety/ Lab 1	Syllabus, safety, glassware/drawer check-in/Beanium Intro to Data
Lab #2	Significant Figures (SF)/Volumetric Glassware Measurements with Pop/Soda
Asynch #1	Excel Statistics, Chem and Lab Safety Labsters
Lab #3	Hard Water Analysis
Asynch #2	Solution Prep and Titration Labsters
Lab #4	Vitamin C Analysis via Titration
Asynch #3	Formal Report Writing and Pipetting Labster
Lab #5	Mouthwash Analysis via Titration
Asynch #4	Formal Report Peer Review
Lab #6	Sports Drink Analysis in Lab, Excel Lab Data Graph for Homework
Asynch #5	Spectrophotometer Labster, Final Draft Formal Report Update
Lab #7	Aspirin Analysis in Lab, Excel Lab Data Graph for Homework
Check Out	Wrap up semester, attendance especially required to earn course grade

**Semester calendar schedule of laboratory on the next several pages**

### Tentative Semester Schedule of Chem 111 Laboratory, Spring 2023

Month	Monday	Tuesday	Wednesday	Thursday	Friday
<b>Jan 2023</b>	<b>16</b> <b>MLK Day</b> <b>No classes</b> <b>today</b>	<b>17</b> <b>Semester</b> <b>Starts</b>	<b>18</b>	<b>19</b> ALL STUDENTS: Safety/Lab 1 Beanium	<b>20</b> ALL STUDENTS: Safety/Lab 1 Beanium
	<b>23</b>	<b>24</b>	<b>25</b>	<b>26</b> Group A: Lab 2 Volumetric Glassware  Group B: Asynch 1	<b>27</b> Group A: Lab 2 Volumetric Glassware  Group B: Asynch 1
	<b>30</b>	<b>31</b>	<b>1</b>	<b>2</b> Group B: Lab 2 Volumetric Glassware  Group A: Asynch 1	<b>3</b> Group B: Lab 2 Volumetric Glassware  Group A: Asynch 1
<b>Feb 2023</b>	<b>6</b>	<b>7</b>	<b>8</b>	<b>9</b> Group A: Lab 3 Hard Water  Group B: Asynch 2	<b>10</b> Group A: Lab 3 Hard Water  Group B: Asynch 2
	<b>13</b>	<b>14</b>	<b>15</b>	<b>16</b> Group B: Lab 3 Hard Water  Group A: Asynch 2	<b>17</b> Group B: Lab 3 Hard Water  Group A: Asynch 2
	<b>20</b>	<b>21</b>	<b>22</b>	<b>23</b> Group A: Lab 4 Vitamin C  Group B: Lab 4 Vitamin C	<b>24</b> Group A: Lab 4 Vitamin C  Group B: Lab 4 Vitamin C
	<b>27</b>	<b>28</b>	<b>1</b>	<b>2</b> <b>Make Up Day</b> <b>Labs 1-4</b>  Group A: Asynch 3  Group B: Asynch 3	<b>3</b> <b>Make Up Day</b> <b>Labs 1-4</b>  Group A: Asynch 3  Group B: Asynch 3
<b>Mar 2023</b>	<b>6</b> <b>Spring Break</b> <b>No classes</b> <b>today</b>	<b>7</b> <b>Spring Break</b> <b>No classes</b> <b>today</b>	<b>8</b> <b>Spring Break</b> <b>No classes</b> <b>today</b>	<b>9</b> <b>Spring Break</b> <b>No classes</b> <b>today</b>	<b>10</b> <b>Spring Break</b> <b>No classes</b> <b>today</b>

Month	Monday	Tuesday	Wednesday	Thursday	Friday
Mar 2023	13	14	15	16 Group A: Lab 5 Mouthwash  Group B: Asynch 4	17 Group A: Lab 5 Mouthwash  Group B: Asynch 4
	20	21	22	23 Group B: Lab 5 Mouthwash  Group A: Asynch 4	24 Group B: Lab 5 Mouthwash  Group A: Asynch 4
	27	28	29	30 Group A: Lab 6 Sports Drink  Group B: Asynch 5	31 Group A: Lab 6 Sports Drink  Group B: Asynch 5
April 2023	3	4	5	6 <b>Easter Holiday</b> <b>No classes today</b>	7 <b>Easter Holiday</b> <b>No classes today</b>
	10	11	12	13 Group B: Lab 6 Sports Drink  Group A: Asynch 5	14 Group B: Lab 6 Sports Drink  Group A: Asynch 5
	17	18	19	20 <b>All students in lab!!!</b> Group A: Lab 7 Aspirin  Group B: Lab 7 Aspirin	21 <b>All students in lab!!!</b> Group A: Lab 7 Aspirin  Group B: Lab 7 Aspirin
	24	25	26	27 <b>Make Up Day</b> <b>Labs 5-7</b>  Checkout/Research Opportunities	28 <b>Semester Ends</b> <b>Make Up Day</b> <b>Labs 5-7</b>  Checkout/Research Opportunities

**Semester weekly schedule of laboratory and course work on next several pages for both groups A and B**

**Group A Schedule**

<b>Week and Class Dates</b>	<b>Weekly Activity</b>	<b>Material Due</b>
<p><b>Week 1</b> Jan 19<sup>th</sup> – 20<sup>th</sup></p>	<p><b>Synchronous – Lab 1</b> Safety, Lab # 1 Bermanium</p>	<p><b>Due By Week 2</b></p> <ol style="list-style-type: none"> <li>1. Read syllabus in Sakai (<i>Syllabus</i>) and explore site.</li> <li>2. Buy required items.</li> <li>3. Watch lecture on Lab #2 Volumetric Measurements Lab (<i>Panopto</i>)</li> <li>4. Read PDF lab manual in Sakai for Volumetric Measurements experiment (<i>Resources</i>)</li> <li>5. Watch JoVE videos on Lab Glassware and Lab Techniques</li> <li>6. Try Labster demo on Sakai (<i>Labster menu in Sakai</i>)</li> <li>7. Write necessary info in lab notebook, will be checked at start of class</li> </ol> <p><b>Due By Week 3</b></p> <ol style="list-style-type: none"> <li>1. <i>Submit Bermanium Lab notebook pages/data sheet to Sakai (Assignments)</i></li> </ol>
<p><b>Week 2</b> Jan 26<sup>th</sup> – 27<sup>th</sup></p>	<p><b>Synchronous – Lab 2</b></p> <ol style="list-style-type: none"> <li>1. Complete Lab #2 Volumetric Glassware</li> <li>2. Collect classroom data</li> <li>3. If you missed lab last week you must do Lab #1 AND Lab #2!!</li> </ol>	<p><b>Due By Week 3</b></p> <ol style="list-style-type: none"> <li>1. <i>Submit Bermanium Lab notebook pages/data sheet to Sakai (Assignments)</i></li> <li>2. <i>Syllabus, Course Policies, Safety quiz in Sakai (Tests and Quizzes)</i></li> </ol> <p><b>Due by Week 4</b></p> <ol style="list-style-type: none"> <li>1. <i>Submit Volumetric Glassware notebook pages to Sakai (Assignments) and Lab #1 data sheet if applicable.</i></li> </ol>

<p><b>Week 3</b> Feb 2<sup>nd</sup> – 3<sup>rd</sup></p>	<p><b>Asynchronous 1</b></p> <ol style="list-style-type: none"> <li>1. Complete Lab Safety Labster</li> <li>2. Complete Chemical Safety Labster,</li> <li>3. Excel Statistics activity</li> </ol>	<p><b>Due by Week 4</b></p> <ol style="list-style-type: none"> <li>1. Both Labster activities in Sakai (<i>Labster menu in Sakai</i>)</li> <li>2. <a href="#">Submit Excel Stats File to Sakai (Assignments)</a></li> <li>3. Watch lecture on Lab #3 Hard Water lab (<i>Panopto</i>)</li> <li>4. Read PDF lab manual in Sakai for Hard Water experiment (<i>Resources</i>)</li> <li>5. Write necessary info in lab notebook, will be checked at start of class.</li> </ol>
<p><b>Week 4</b> Feb 9<sup>th</sup> – 10<sup>th</sup></p>	<p><b>Synchronous – Lab 3</b></p> <ol style="list-style-type: none"> <li>1. Complete Lab #3 Hard Water</li> <li>2. Collect classroom data</li> </ol>	<p><b>Due by Week 5</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Beanium quiz in Sakai (Tests &amp; Quizzes)</a></li> </ol> <p><b>Due by Week 6</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Hard Water Lab notebook pages to Sakai (Assignments)</a></li> </ol>
<p><b>Week 5</b> Feb 16<sup>th</sup> – Feb 17<sup>th</sup></p>	<p><b>Asynchronous 2</b></p> <ol style="list-style-type: none"> <li>1. Complete Solution Prep Labster</li> <li>2. Complete Titration Labster</li> </ol>	<p><b>Due by Week 6</b></p> <ol style="list-style-type: none"> <li>1. Both Labster activities in Sakai (<i>Labster menu in Sakai</i>)</li> <li>2. Watch lecture on Lab #4 Vitamin C (<i>Panopto</i>)</li> <li>3. Read PDF lab manual in Sakai for Vitamin C experiment (<i>Resources</i>)</li> <li>4. Write necessary info in lab notebook, will be checked at start of class.</li> <li>5. Watch JoVE videos on Solution Preparation and Stoichiometry and Titrations.</li> </ol>
<p><b>Week 6</b> Feb 23<sup>rd</sup> - Feb 24<sup>th</sup></p>	<p><b>Synchronous – Lab 4</b></p> <ol style="list-style-type: none"> <li>1. Complete Lab #4 Vitamin C</li> <li>2. Collect classroom data</li> </ol>	<p><b>Due By Week 8:</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Vitamin C Lab notebook pages to Sakai (Assignments)</a></li> <li>2. You are asynchronous next week typing a formal lab report. All information is posted in Sakai (<i>Formal Lab Report</i>) if you want to get a head start.</li> </ol>

<p><b>Week 7</b> Mar 2<sup>nd</sup> – Mar 3<sup>rd</sup></p>	<p><b>Synchronous – Make up</b></p> <ol style="list-style-type: none"> <li>1. Make up lab 1-4 if missed earlier in the semester.</li> </ol> <p><b>Asynchronous 3</b></p> <ol style="list-style-type: none"> <li>1. Learn how to type a formal lab report</li> <li>2. Type a rough draft of Hard Water Analysis lab formal lab report.</li> <li>3. Complete Pipetting Labster</li> </ol>	<p><b>Due By Next Week 8</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Formal lab report rough draft on Hard Water Lab, Sakai (Assignments)</a></li> <li>2. Pipetting Labster in Sakai (<i>Labster menu in Sakai</i>)</li> <li>3. <a href="#">Hard Water quiz in Sakai (Tests &amp; Quizzes)</a></li> <li>4. Watch lecture on Lab #5 Mouthwash (<i>Panopto</i>)</li> <li>5. Read PDF lab manual in Sakai for Mouthwash experiment (<i>Resources</i>)</li> <li>6. Write necessary info in lab notebook, will be checked at start of class.</li> </ol>
<p><b>Spring Break</b> Mar 6<sup>th</sup> – Mar 10<sup>th</sup></p>	<p><b>No Classes</b></p>	<p><b>No Classes</b></p>
<p><b>Week 8</b> Mar 16<sup>th</sup> – Mar 17<sup>th</sup></p>	<p><b>Synchronous – Lab 5</b></p> <ol style="list-style-type: none"> <li>1. Complete Lab #5 Mouthwash Analysis</li> <li>2. Collect classroom data</li> </ol>	<p><b>Due By Week 10</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Mouthwash Lab notebook pages to Sakai (Assignments)</a></li> </ol>
<p><b>Week 9</b> Mar 23<sup>rd</sup> – Mar 24<sup>th</sup></p>	<p><b>Asynchronous 4</b></p> <ol style="list-style-type: none"> <li>1. Learn how to peer review</li> <li>2. Complete peer reviews in Sakai.</li> </ol>	<p><b>Due By Week 10:</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Complete peer reviews in Sakai (Assignments)</a></li> <li>2. Watch lecture on Lab #6 Sports Drink (<i>Panopto</i>)</li> <li>3. Read PDF lab manual in Sakai for Sports Drink experiment (<i>Resources</i>)</li> <li>4. Write necessary info in lab notebook, will be checked at start of class.</li> <li>5. <a href="#">Titration &amp; Stoichiometry quiz in Sakai (Tests &amp; Quizzes)</a></li> <li>6. Watch JoVE videos on Beers Law and Spectrophotometry</li> </ol>
<p><b>Week 10</b> Mar 30<sup>th</sup> – Mar 31<sup>st</sup></p>	<p><b>Synchronous – Lab 6</b></p> <ol style="list-style-type: none"> <li>1. Complete Lab #6 Sports Drink Analysis</li> <li>2. Collect classroom data</li> <li>3. Recreate calibration curve from lab 6 in excel</li> </ol>	<p><b>Due By Week 12:</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Sports Drink Lab notebook pages to Sakai (Assignments)</a></li> <li>2. <a href="#">Submit Excel of Sports Drink Standard Curve to Sakai (Assignments)</a></li> </ol>

<b>Easter Holiday</b> Apr 6 <sup>th</sup> – Apr 7 <sup>th</sup>	<b>No Classes</b>	<b>No Classes</b>
<b>Week 11</b> Apr 13 <sup>th</sup> – Apr 14 <sup>th</sup>	<b>Asynchronous 5</b> <ol style="list-style-type: none"> <li>1. Spectrophotometer Labster</li> <li>2. Review peer feedback &amp; update Formal Report.</li> </ol>	<b>Due By Week 12:</b> <ol style="list-style-type: none"> <li>1. Labster activity in Sakai (<i>Labster menu in Sakai</i>)</li> <li>2. <a href="#">Revise, then submit final draft of Formal Report to Sakai (Assignments)</a></li> <li>3. <a href="#">Spectrophotometric analysis quiz in Sakai (Tests &amp; Quizzes)</a></li> <li>4. Watch lecture on Lab #7 Aspirin (<i>Panopto</i>)</li> <li>5. Read PDF lab manual in Sakai for Aspirin experiment (<i>Resources</i>)</li> <li>6. Write necessary info in lab notebook, will be checked at start of class.</li> </ol>
<b>Week 12</b> Apr 20 <sup>th</sup> – Apr 21 <sup>st</sup>	<b>Synchronous – Lab 7</b> <ol style="list-style-type: none"> <li>1. Complete Lab #7 Aspirin Analysis</li> <li>2. Collect classroom data</li> <li>3. Recreate calibration curve from lab 7 in excel</li> </ol>	<b>Due By Week 13</b> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Aspirin Lab notebook pages to Sakai (Assignments)</a></li> <li>2. <a href="#">Submit Excel of Aspirin Standard Curve to Sakai (Assignments)</a></li> <li>3. <a href="#">Cumulative topics quiz in Sakai (Tests &amp; Quizzes)</a></li> <li>4. Check Gradebook in Sakai and ask questions to lab coordinator.</li> </ol>
<b>Week 13</b> Apr 27 <sup>th</sup> – Apr 28 <sup>th</sup>	<b>Checkout/ Makeup Day</b> <ol style="list-style-type: none"> <li>1. Clean lab glassware</li> <li>2. Ask last minute grading questions</li> <li>3. Chat about undergraduate research opportunities, etc.</li> </ol>	<b>Review any grade issues with instructor.</b>

## Group B Schedule

Week and Class Dates	Weekly Activity	Material Due
<b>Week 1</b> Jan 19 <sup>th</sup> – 20 <sup>th</sup>	<b>Synchronous – Lab 1</b> Safety, Lab # 1 Bermanium	<b>Due By Week 2</b> <ol style="list-style-type: none"> <li>1. Read syllabus in Sakai (<i>Syllabus</i>) and explore site.</li> <li>2. Buy required items.</li> <li>3. Try Labster demo on Sakai (<i>Labster</i>)</li> </ol> <b>Due By Week 3</b> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Bermanium Lab notebook pages/data sheet to Sakai (Assignments)</a></li> </ol>
<b>Week 2</b> Jan 26 <sup>th</sup> – 27 <sup>th</sup>	<b>Asynchronous 1</b> <ol style="list-style-type: none"> <li>1. Complete Lab Safety Labster</li> <li>2. Complete Chemical Safety Labster,</li> <li>3. Excel Statistics activity</li> </ol>	<b>Due By Week 3</b> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Bermanium Lab notebook pages/data sheet to Sakai (Assignments)</a></li> <li>2. <a href="#">Syllabus, Course Policies, Safety quiz in Sakai (Tests and Quizzes)</a></li> <li>3. Watch lecture on Lab #2 Volumetric Measurements Lab (<i>Panopto</i>)</li> <li>4. Read PDF lab manual in Sakai for Volumetric Measurements experiment (<i>Resources</i>)</li> <li>5. Watch JoVE videos on Lab Glassware and Lab Techniques</li> <li>6. Write necessary info in lab notebook, will be checked at start of class</li> </ol> <b>Due by Week 4</b> <ol style="list-style-type: none"> <li>1. Both Labster activities in Sakai (<i>Labster menu in Sakai</i>)</li> <li>2. <a href="#">Submit Excel Stats File to Sakai (Assignments)</a></li> </ol>



<b>Week 3</b> Feb 2 <sup>nd</sup> – 3 <sup>rd</sup>	<b>Synchronous – Lab 2</b> <ol style="list-style-type: none"> <li>1. Complete Lab #2 Volumetric Glassware</li> <li>2. Collect classroom data</li> <li>3. If you missed lab last week you must do Lab #1 AND Lab #2!!</li> </ol>	<b>Due by Week 4</b> <ol style="list-style-type: none"> <li>1. Submit Volumetric Glassware notebook pages to Sakai (<i>Assignments</i>) and Lab #1 data sheet if applicable.</li> </ol>
<b>Week 4</b> Feb 9 <sup>th</sup> – 10 <sup>th</sup>	<b>Asynchronous 2</b> <ol style="list-style-type: none"> <li>1. Complete Solution Prep Labster</li> <li>2. Complete Titration Labster</li> </ol>	<b>Due by Week 5</b> <ol style="list-style-type: none"> <li>1. Watch lecture on Lab #3 Hard Water lab (<i>Panopto</i>)</li> <li>2. Read PDF lab manual in Sakai for Hard Water experiment (<i>Resources</i>)</li> <li>3. Write necessary info in lab notebook, will be checked at start of class.</li> <li>4. Beanium quiz in Sakai (<i>Tests &amp; Quizzes</i>)</li> </ol> <b>Due by Week 6</b> <ol style="list-style-type: none"> <li>1. Both Labster activities in Sakai (<i>Labster menu in Sakai</i>)</li> </ol>
<b>Week 5</b> Feb 16 <sup>th</sup> – Feb 17 <sup>th</sup>	<b>Synchronous – Lab 3</b> <ol style="list-style-type: none"> <li>1. Complete Lab #3 Hard Water</li> <li>2. Collect classroom data</li> </ol>	<b>Due by Week 6</b> <ol style="list-style-type: none"> <li>1. Submit Hard Water Lab notebook pages to Sakai (<i>Assignments</i>)</li> <li>2. Watch lecture on Lab #4 Vitamin C (<i>Panopto</i>)</li> <li>3. Read PDF lab manual in Sakai for Vitamin C experiment (<i>Resources</i>)</li> <li>4. Write necessary info in lab notebook, will be checked at start of class.</li> <li>5. Watch JoVE videos on Solution Preparation and Stoichiometry and Titrations.</li> </ol>
<b>Week 6</b> Feb 23 <sup>rd</sup> - Feb 24 <sup>th</sup>	<b>Synchronous – Lab 4</b> <ol style="list-style-type: none"> <li>1. Complete Lab #4 Vitamin C</li> <li>2. Collect classroom data</li> </ol>	<b>Due By Week 8:</b> <ol style="list-style-type: none"> <li>1. Submit Vitamin C Lab notebook pages to Sakai (<i>Assignments</i>)</li> <li>2. You are asynchronous next week typing a formal lab report. All information is posted in Sakai (<i>Formal Lab Report</i>) if you want to get a head start.</li> </ol>

<p><b>Week 7</b> Mar 2<sup>nd</sup> – Mar 3<sup>rd</sup></p>	<p><b>Synchronous – Make up</b></p> <ol style="list-style-type: none"> <li>1. Make up lab 1-4 if missed earlier in the semester.</li> </ol> <p><b>Asynchronous 3</b></p> <ol style="list-style-type: none"> <li>1. Learn how to type a formal lab report</li> <li>2. Type a rough draft of Hard Water Analysis lab formal lab report.</li> <li>3. Complete Pipetting Labster</li> </ol>	<p><b>Due By Week 8</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Formal lab report rough draft on Hard Water Lab, Sakai (Assignments)</a></li> <li>2. Pipetting Labster in Sakai (<i>Labster menu in Sakai</i>)</li> <li>3. <a href="#">Hard Water quiz in Sakai (Tests &amp; Quizzes)</a></li> </ol>
<p><b>Spring Break</b> Mar 6<sup>th</sup> – Mar 10<sup>th</sup></p>	<p><b>No Classes</b></p>	<p><b>No Classes</b></p>
<p><b>Week 8</b> Mar 16<sup>th</sup> – Mar 17<sup>th</sup></p>	<p><b>Asynchronous 4</b></p> <ol style="list-style-type: none"> <li>1. Learn how to peer review</li> <li>2. Complete peer reviews in Sakai.</li> </ol>	<p><b>Due By Week 9</b></p> <ol style="list-style-type: none"> <li>1. Watch lecture on Lab #5 Mouthwash (<i>Panopto</i>)</li> <li>2. Read PDF lab manual in Sakai for Mouthwash experiment (<i>Resources</i>)</li> <li>3. Write necessary info in lab notebook, will be checked at start of class.</li> </ol> <p><b>Due By Week 10:</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Complete peer reviews in Sakai (Assignments)</a></li> </ol>
<p><b>Week 9</b> Mar 23<sup>rd</sup> – Mar 24<sup>th</sup></p>	<p><b>Synchronous – Lab 5</b></p> <ol style="list-style-type: none"> <li>1. Complete Lab #5 Mouthwash Analysis</li> <li>2. Collect classroom data</li> </ol>	<p><b>Due By Week 10:</b></p> <ol style="list-style-type: none"> <li>1. <a href="#">Submit Mouthwash Lab notebook pages to Sakai (Assignments)</a></li> <li>2. Watch lecture on Lab #6 Sports Drink (<i>Panopto</i>)</li> <li>3. Read PDF lab manual in Sakai for Sports Drink experiment (<i>Resources</i>)</li> <li>4. Write necessary info in lab notebook, will be checked at start of class.</li> <li>5. <a href="#">Titration &amp; Stoichiometry quiz in Sakai (Tests &amp; Quizzes)</a></li> <li>6. Watch JoVE videos on Beers Law and Spectrophotometry</li> </ol>
<p><b>Week 10</b> Mar 30<sup>th</sup> – Mar 31<sup>st</sup></p>	<p><b>Asynchronous 5</b></p> <ol style="list-style-type: none"> <li>1. Spectrophotometer Labster</li> <li>2. Review peer feedback &amp; update Formal Report.</li> </ol>	<p><b>Due By Week 12:</b></p> <ol style="list-style-type: none"> <li>1. Labster activity in Sakai (<i>Labster menu in Sakai</i>)</li> <li>2. <a href="#">Revise, then submit final draft of Formal Report to Sakai(Assignments)</a></li> </ol>

<b>Easter Holiday</b> Apr 6 <sup>th</sup> – Apr 7 <sup>th</sup>	<b>No Classes</b>	<b>No Classes</b>
<b>Week 11</b> Apr 13 <sup>th</sup> – Apr 14 <sup>th</sup>	<b>Synchronous – Lab 6</b> <ol style="list-style-type: none"> <li>1. Complete Lab #6 Sports Drink Analysis</li> <li>2. Collect classroom data</li> <li>3. Recreate calibration curve from lab 6 in excel</li> </ol>	<b>Due By Week 12:</b> <ol style="list-style-type: none"> <li>1. Submit Sports Drink Lab notebook pages to Sakai (<i>Assignments</i>)</li> <li>2. Submit Excel of Sports Drink Standard Curve to Sakai (<i>Assignments</i>)</li> <li>3. Spectrophotometric analysis quiz in Sakai (<i>Tests &amp; Quizzes</i>)</li> <li>4. Watch lecture on Lab #7 Aspirin (<i>Panopto</i>)</li> <li>5. Read PDF lab manual in Sakai for Aspirin experiment (<i>Resources</i>)</li> <li>6. Write necessary info in lab notebook, will be checked at start of class.</li> </ol>
<b>Week 12</b> Apr 20 <sup>th</sup> – Apr 21 <sup>st</sup>	<b>Synchronous – Lab 7</b> <ol style="list-style-type: none"> <li>1. Complete Lab #7 Aspirin Analysis</li> <li>2. Collect classroom data</li> <li>3. Recreate calibration curve from lab 7 in excel</li> </ol>	<b>Due By Week 13</b> <ol style="list-style-type: none"> <li>1. Submit Aspirin Lab notebook pages to Sakai (<i>Assignments</i>)</li> <li>2. Submit Excel of Aspirin Standard Curve to Sakai (<i>Assignments</i>)</li> <li>3. Cumulative topics quiz in Sakai (<i>Tests &amp; Quizzes</i>)</li> <li>4. Check Gradebook in Sakai and ask questions to lab coordinator.</li> </ol>
<b>Week 13</b> Apr 27 <sup>th</sup> – Apr 28 <sup>th</sup>	<b>Checkout/ Makeup Day</b> <ol style="list-style-type: none"> <li>1. Clean lab glassware</li> <li>2. Ask last minute grading questions</li> <li>3. Chat about undergraduate research opportunities, etc.</li> </ol>	<b>Review any grade issues with instructor.</b>